

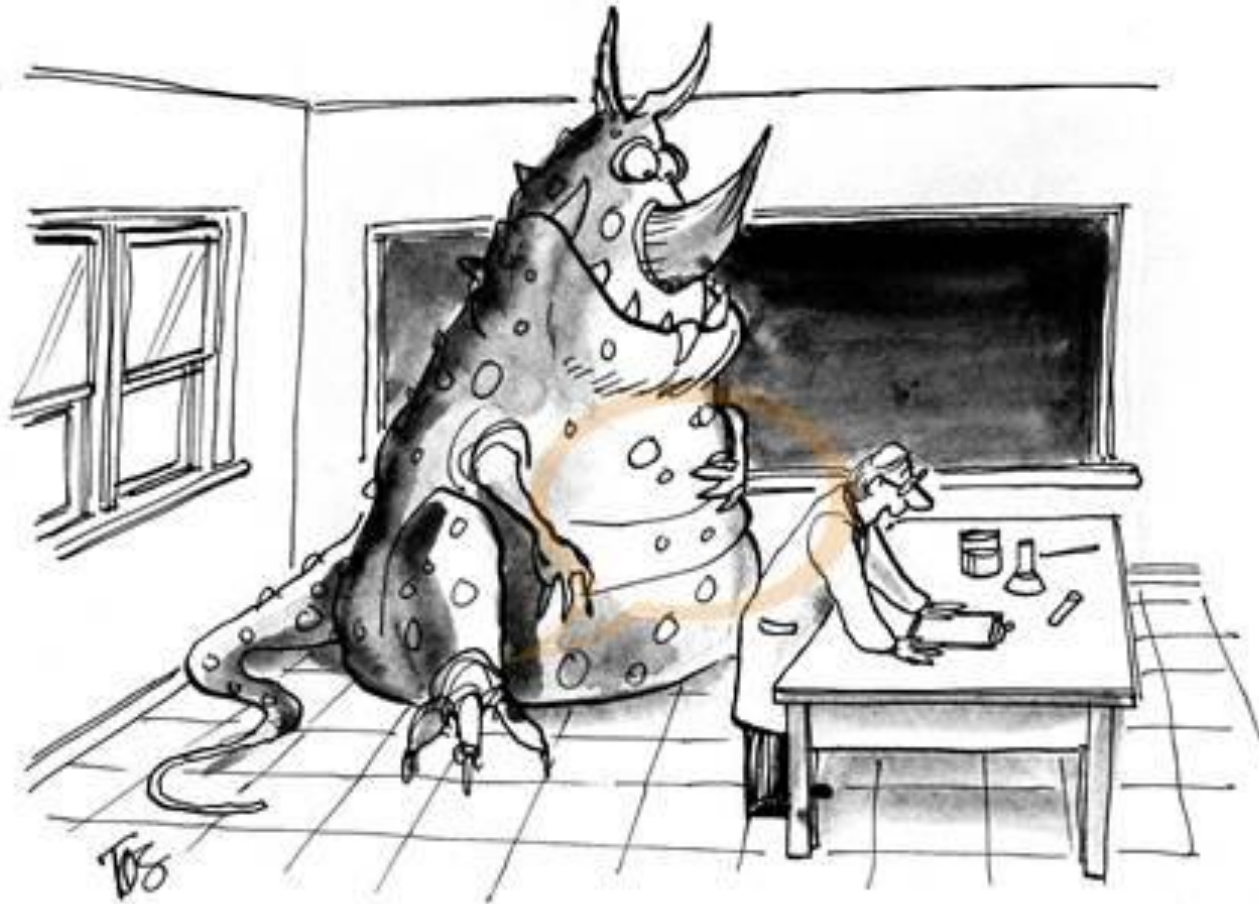
# **NSWBI Technical Webinar #2**

## **Dumps, Landfills, Emerging Contaminants and Impacts on Landfill Leachate Quality due to Increased Organics Diversion:**

### **What do we expect and what do we know?**

Monica Danon-Schaffer, PhD., P.Eng.

# Dumps/Landfills End of Life (EOL) Management



"Son of a gun, you're right, there were some harmful compounds in those chemicals we dumped."

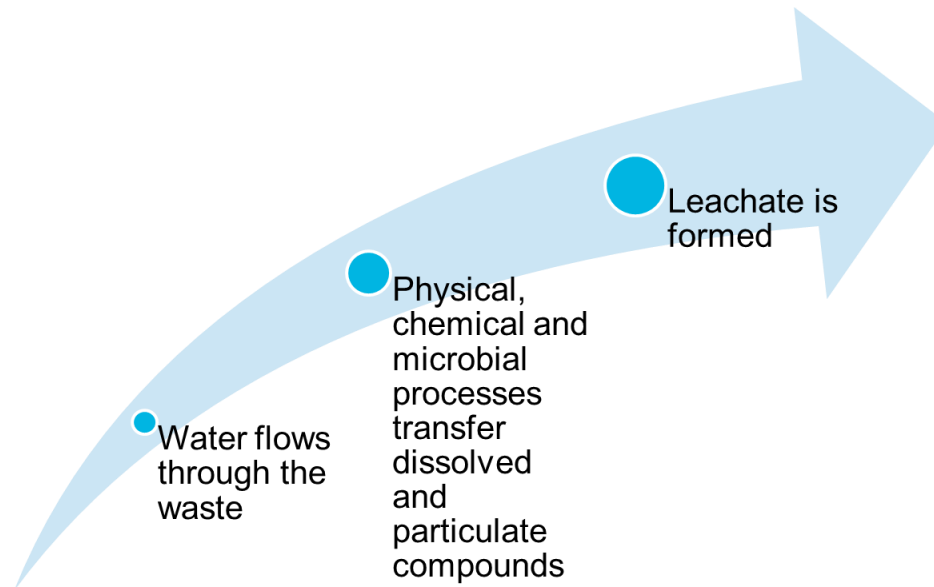
# Today's Discussion

- Background on Landfill Leachate Generation and Characteristics
- Leachate Treatment
- Leachate Quality and Increased Organics Diversion
- Case Studies
- Emerging Contaminants in Landfills
- Unique Challenges with Northern dumpsites/landfills
- What Comes Next?

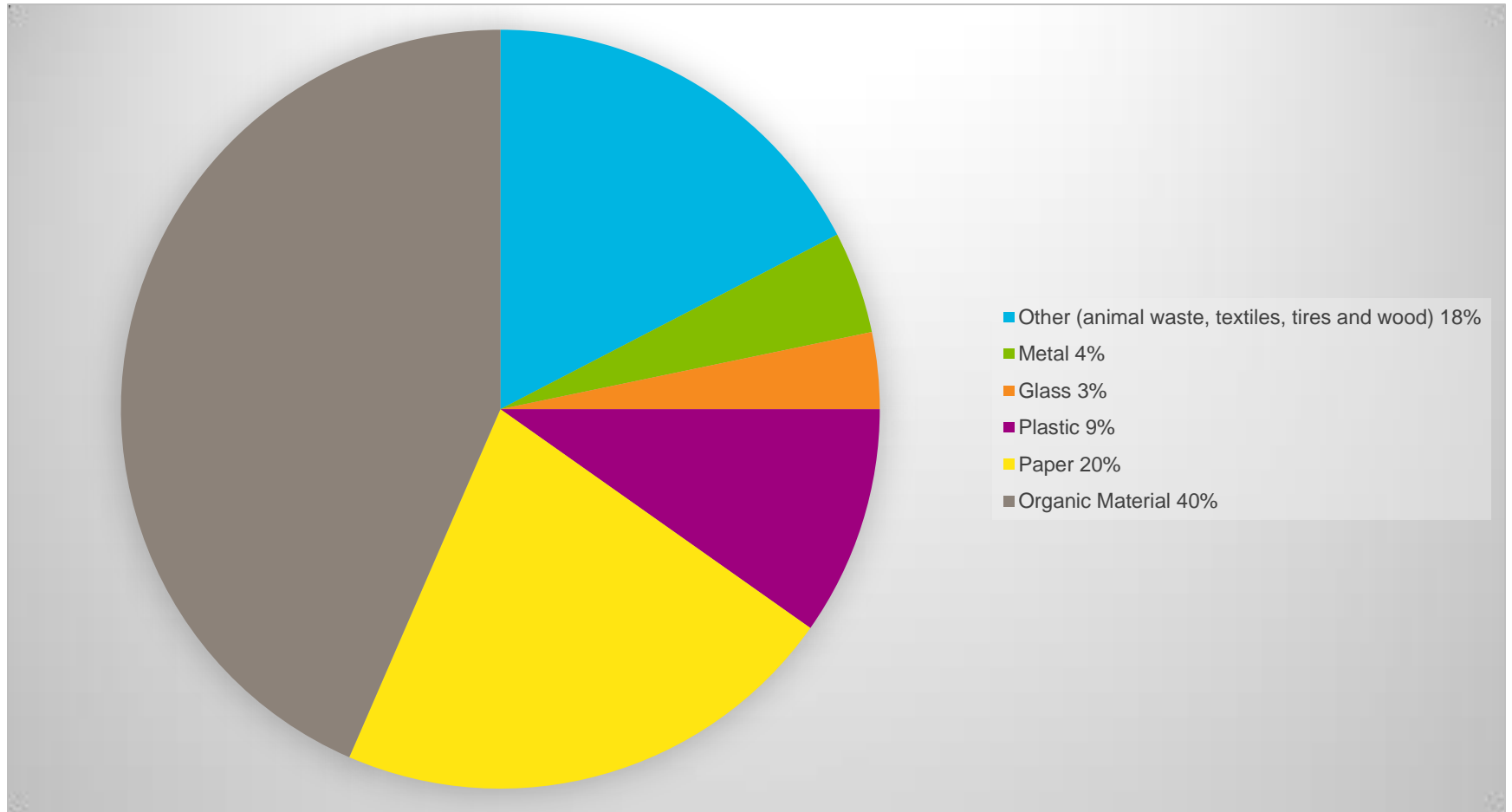
# **Background on Landfill Leachate Generation and Characteristics**

# What is Leachate?

- Formed when water *percolates* through a substance and acquires soluble materials from that substance
  - Ex. Water through coffee grounds
- Waste *composition* affects composition of leachate
- Dark brown/black, strong odour



# Composition of Waste in Canada



Statistics Canada 2005

# Four Main Contaminant Groups

## Dissolved organic matter

- Alcohols, acids, aldehydes, short chain sugars, and more stable organics

## Inorganic macrocomponents

- Cations and ions including: calcium, magnesium, manganese, sodium, potassium, chloride, sulphate, iron, ammonia, and carbonates

## Metals

- Dissolved metals including: lead and mercury

## Organic compounds

- Aromatic hydrocarbons, phenols, halogenated organics (i.e. PCBs, PBDEs, PFCs, dioxins), pesticides, solvents, and plasticizers

# Four Parameters for Characterization

## pH

- indicates level of acidity of the liquid

## Specific conductance

- ability to conduct electricity

## Chemical Oxygen Demand (COD)

- indicator of total amount of organic matter in leachate

## Biochemical Oxygen Demand (BOD)

- measure of amount of biodegradable organic matter in leachate

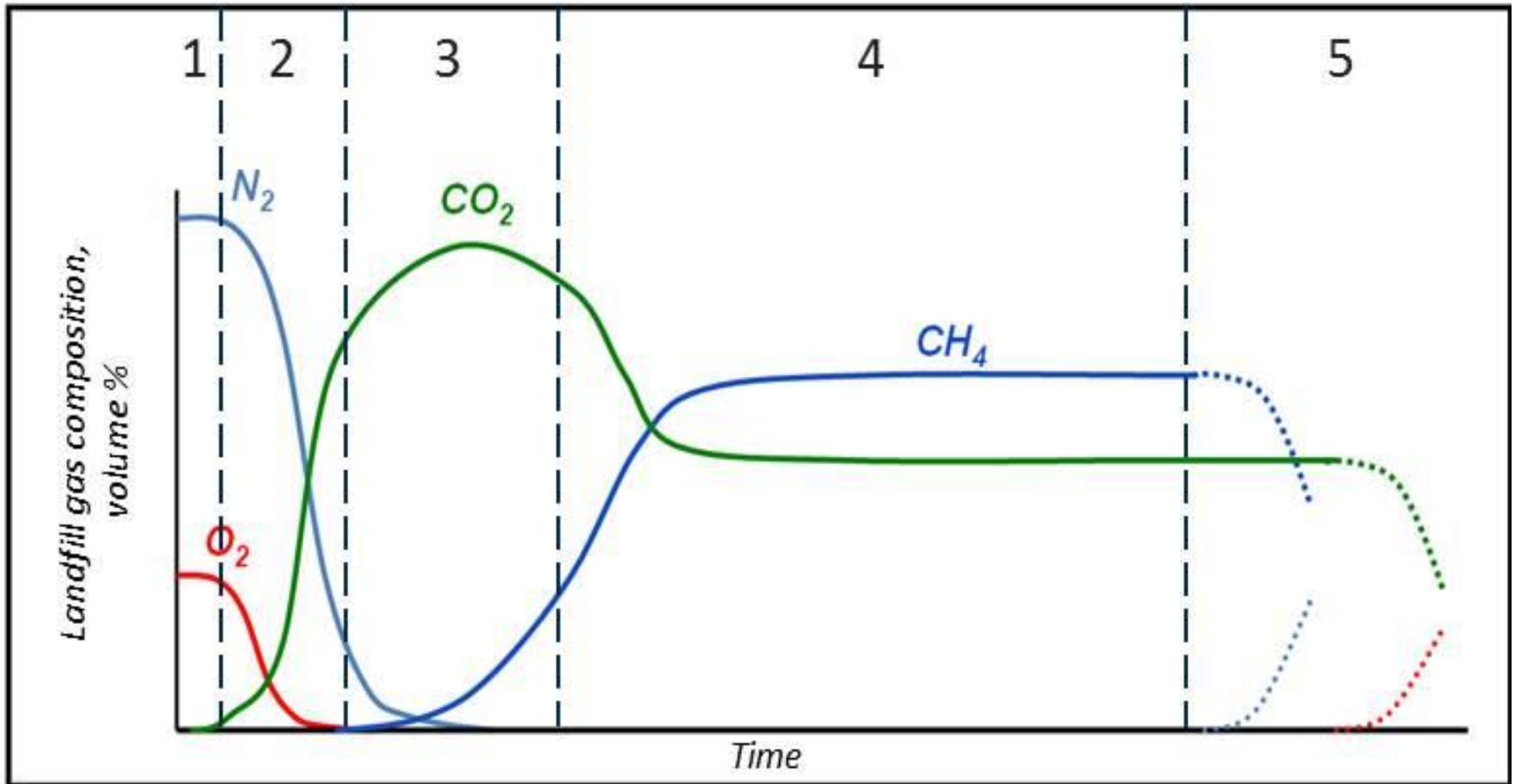


| Parameter  | Units              | Municipal Solid Waste   | Industrial Waste                 |
|--|--------------------|-------------------------|----------------------------------|
|  |                    | Range                   | Range                            |
| pH   | -                  | 1.9 – 13.0              | 5.8 – 12.6                       |
| Electrical conductivity  | uS/cm <sup>2</sup> | 3 – 92,900              | 275 – 65,700                     |
| Total dissolved solids (TDS)   | mg/L               | 1 – 520,000             | 136 – 52,200                     |
| Total suspended solids (TSS)   | mg/L               | 3 – 21,940              | 01 – 4,400                       |
| BOD  | mg/L               | 2 – 561,000             | b.d.l. (2) <sup>a</sup> – 14,400 |
| COD  | mg/L               | 10 – 83,000             | b.d.l. (2) – 25,500              |
| Total alkalinity (CaCO <sub>3</sub> )  | mg/L               | 1 – 30,900              | 61 – 4,078                       |
| Sulfate (SO <sub>4</sub> <sup>2-</sup> )   | mg/L               | 1 – 64,703              | 1 – 2,130                        |
| Chloride (Cl <sup>-</sup> )  | mg/L               | 11.9 – 30,200           | 5 – 32,000                       |
| Phosphate (PO <sub>4</sub> <sup>3-</sup> )                                       | mg/L               | 0.01 – 885              | b.d.l. (0.01)                    |
| Ammonia (NH <sub>3</sub> )   | mg/L               | 0.1 – 1,900             | 0.1 – 417                        |
| Nitrate + nitrite (NO <sub>2</sub> <sup>-</sup> + NO <sub>3</sub> <sup>-</sup> ) | mg/L               | 0.01 – 165              | b.d.l. (0.05) – 91               |
| Total Kjeldahl nitrogen (TKN)  | mg/L               | b.d.l. (0.2) – 12,000   | 0.2 – 476                        |
| Sodium (Na <sup>+</sup> )  | mg/L               | b.d.l. (1) – 13,000     | 13 – 17,200                      |
| Cadmium (Cd)   | mg/L               | b.d.l. (0.0002) – 255   | b.d.l. (0.0002) – 0.058          |
| Calcium (Ca)   | mg/L               | b.d.l. (0.5) – 3,810    | 30 – 8.820                       |
| Iron (Fe)  | mg/L               | 0.005 - 7260            | 0.02 – 207                       |
| Lead (Pb)  | mg/L               | b.d.l. (0.0001) – 0.674 | b.d.l. (0.0001) – 0.166          |
| Magnesium (Mg)   | mg/L               | b.d.l. (0.1) – 7,170    | 4 – 6,500                        |
| Manganese (Mn)   | mg/L               | b.d.l. (0.001) - 252    | b.d.l. (0.001) - 64              |
| Nickel (Ni)  | mg/L               | b.d.l. (0.0002) – 46.8  | 0.001 – 8.3                      |
| Acetone  | mg/L               | b.d.l. (0.06) – 2,000   | n.d. <sup>b</sup>                |
| Benzene  | mg/L               | b.d.l. (0.001) - 10     | b.d.l. (0.001) - 39              |

# Example of Leachate Quality

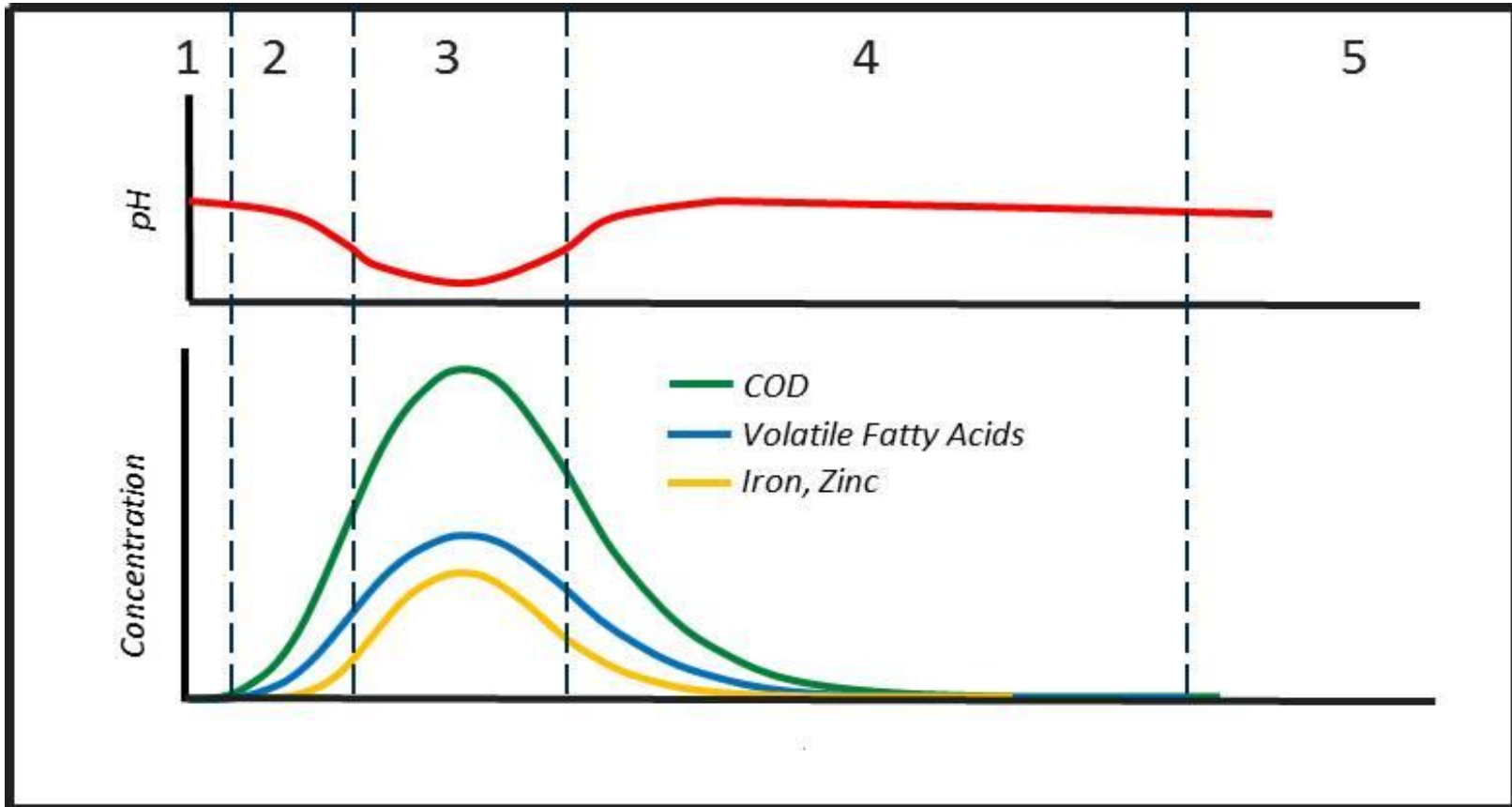
| Parameter                      |      | MSW Leachate |        | Edmonton Leachate (1) |       | Regina Leachate (2) |      | East Prince (NS) Leachate (3) |      |
|--------------------------------|------|--------------|--------|-----------------------|-------|---------------------|------|-------------------------------|------|
|                                | Unit | Min          | Max    | Min                   | Max   | Min                 | Max  | Min                           | Max  |
| pH                             |      | 1.9          | 13     | 7.2                   | 12.6  | 7.59                | 7.92 | 7.12                          | 8.92 |
| BOD5                           | mg/l | 2            | 561000 | 61                    | 20900 |                     |      | 4                             | 52   |
| COD                            | mg/l | 10           | 83000  | 140                   | 35000 | 310                 | 370  | 376                           | 743  |
| Alkalinity as CaCO3            | mg/l | 1            | 30900  | 910                   | 11000 | 730                 | 1400 |                               |      |
| 1) West Edmonton leachate 2009 |      |              |        |                       |       |                     |      |                               |      |
| 2) Regina - leachate tank 2016 |      |              |        |                       |       |                     |      |                               |      |
| 3) East Prince Feb 2017        |      |              |        |                       |       |                     |      |                               |      |

# Leachate Life Cycle: Changes in gas production over time in a landfill. Decomposition phases 1-5



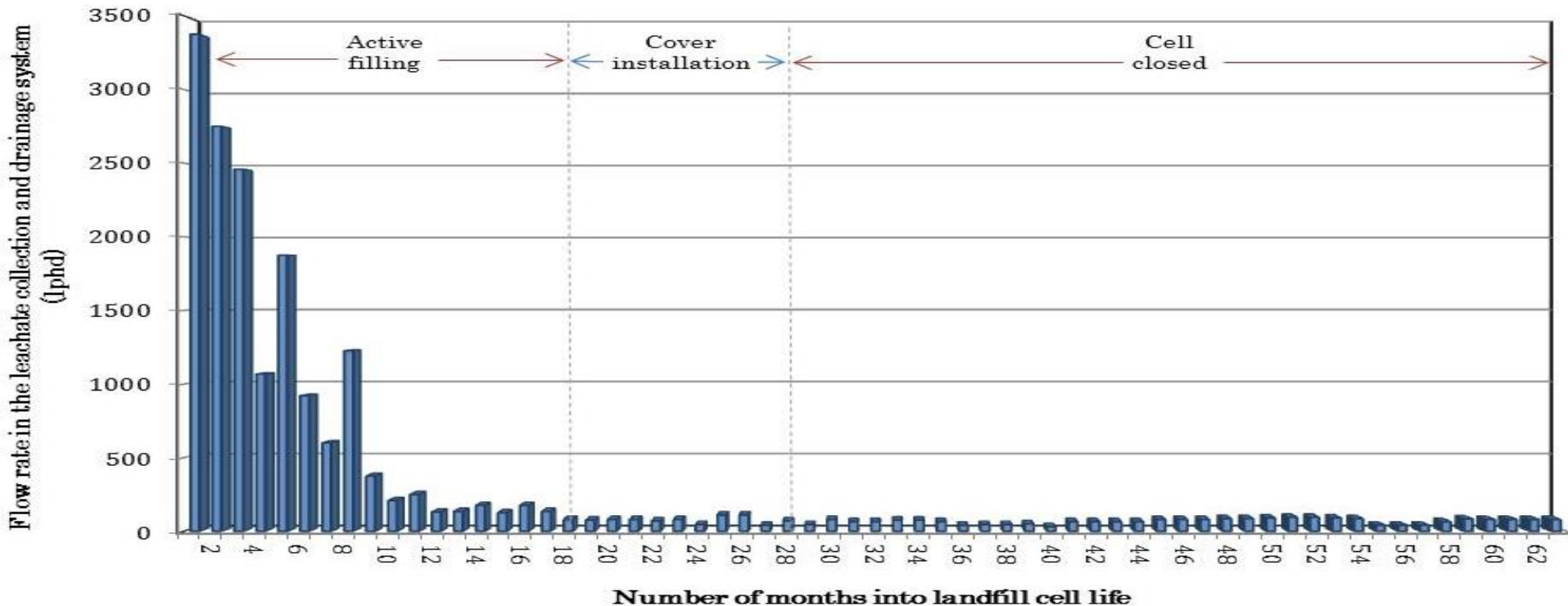
# Leachate Life Cycle: Example of changes in some leachate parameters over time in a landfill.

## Decomposition phases 1-5



# Leachate Production over Time

- Most landfills have a higher leachate production during active operation than post-closure.
- Production rates can be very high during initial operation of a new cell.



## Landfill Operation: Leachate Monitoring

- Monitoring must be carried out through the entire life of the landfill to the end of the post-closure period.
- The leachate monitoring plan should include:
  - leachate levels and quality in the landfill, leachate ponds, and leak detection system;
  - leachate removal and treatment/disposal records;
  - surface and groundwater monitoring; and,
  - record keeping.

*Monitoring helps Operators understand how the collection system is working, if liners are leaking, and the state of decomposition of waste*

# Leachate Collection System (LCS) Clogging

- Most common cause of LCS failure is pipe clogging
- Caused by buildup of material in:
  - Pipes (pipe walls or perforations)
  - Drainage layer (void spaces)
  - Geotextile filter layer (sedimentation)
- Results:
  - Increased head... non-compliance
  - Increase chances of leakage / seepage

# Leachate Collection System Clogging

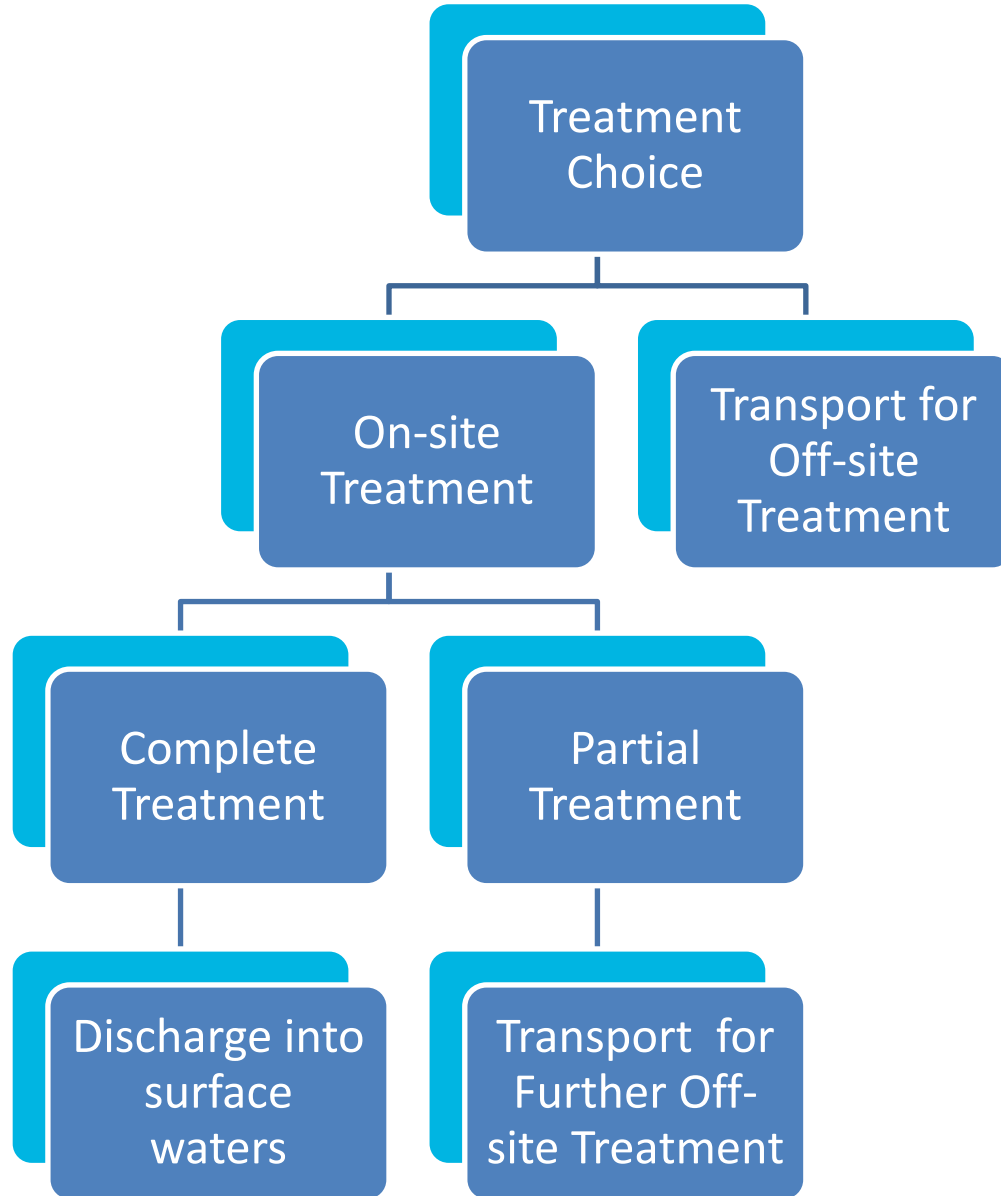
- Where do the solids come from?
  - Particulate matter from waste and soil that leachate flows through
  - Biological material → particles or biofilm formation
  - Precipitation of dissolved chemicals
- Precipitation is strongly affected by pH!
- Most common precipitate in landfills is calcium carbonate





# Leachate Treatment

# Treatment Choice



# Leachate Treatment

– Treatment and disposal options:

- Pretreatment → Municipal sanitary wastewater system
- Transport/disposal to municipal wastewater treatment plant/lagoon
- Deep well disposal
- Treatment and discharge to the environment

# Municipal Wastewater Treatment

- Leachate can be a high strength wastewater and may affect performance of WWTP
  - High organic loads
  - Methanogenic leachate will have high ammonia
  - Toxins
- The treatment plant must have:
  - Capacity to treat leachate volume
  - Ability to treat sludge from leachate
  - Ability to treat contaminants/characteristics of leachate
- Municipal bylaws may require surcharge for disposal or pretreatment

# Municipal Wastewater Treatment

| Parameter            |      | Municipal Leachate |        | Municipal Wastewater |        |        |
|----------------------|------|--------------------|--------|----------------------|--------|--------|
|                      |      | Min                | Max    | Weak                 | Medium | Strong |
|                      | Unit |                    |        |                      |        |        |
| pH                   |      | 1.9                | 13     |                      |        |        |
| BOD5                 | mg/l | 2                  | 561000 | 110                  | 220    | 400    |
| TOC                  | mg/l |                    |        | 80                   | 160    | 290    |
| COD                  | mg/l | 10                 | 83000  | 250                  | 500    | 1000   |
| Total Nitrogen as N  | mg/l |                    |        | 20                   | 40     | 85     |
| Total Phosphate as P | mg/l |                    |        | 4                    | 8      | 15     |
| Alkalinity as CaCO3  | mg/l | 1                  | 30900  | 50                   | 100    | 200    |
| Iron                 | mg/l | 0.005              | 7260   |                      |        |        |

# **Leachate Quality and Increased Organics Diversion**

# Building Better Business Case for Organics Diversion Programs

- Will lead to reduced landfill leachate treatment costs?
- Reduced organics to landfill (~30% of waste stream), reduced airspace consumption and extended landfill life
- Reduced landfill gas generation
- Improved leachate quality
- Social, environmental and financial benefits

## Premise - Removal of Organics

- Improves leachate quality which should:
  - Reduce leachate treatment and/or disposal costs
  - Reduce risk of significant groundwater contamination
- Reduces greenhouse gas generation
- Reduces dependency on foreign oil
- Increases nutrient composition of soil
- Decreases amount of waste to landfill
- Will reduce methane generation



# Case Studies

## Hartland Landfill – Capital Regional District

### –Strategy: phased approach

- 2013: \$20/tonne incentive
- 2014: 20% surcharge at landfill on garbage loads containing kitchen scraps
- 2015: complete ban on kitchen scraps

### –Organics diversion includes: meat, bones, dairy products, grains, eggs, vegetables, fruit, soiled paper products

### –Results:

- Organic waste reduced to 75 kg/person from 120 kg/person in 2010

# Niagara Organics Collection Pilot Study 2013-2014

- Two co-collection scenarios were examined to determine impact on diversion rates/program costs:
  - Co-collection of kitchen organics/recyclables from 657 family households from one community;
  - Another scenario: 606 family households to examine weekly garbage collection/kitchen organics/leaf & yard wastes
- Results:
  - Region could offer weekly collection of recyclables and household kitchen organics for no difference in cost to current recycling collection program
  - Driver finished between 1-3 h earlier than during previous program.

## Happy Valley-Goose Bay Landfill

- No additional staff need for pilot project launched in 2015
- One objective is to mulch organic waste material (i.e. trees, shrubs, logs)
- Increase life of landfill by 25 years

## Newfoundland and Labrador

- Divided into 8 management regions
- Focus on diverting traditional MSW organic materials from residential, industrial, commercial and institutional generators
- Objective to reduce amount of waste entering landfill by 50%

# Emerging Contaminants in Landfills

## What are Emerging Contaminants?

- Any chemical mixture or hazardous material with the *potential* to cause adverse human health, public safety, or environmental effects
- Any synthetic or naturally occurring chemical not typically monitored in the environment but *recently detected* in the environment
- Limited toxicity information available
- Concern with the length of exposure time
- Compounds designated as *candidate* Persistent Organic Pollutants (POPs)

# Emerging Contaminants of Concern

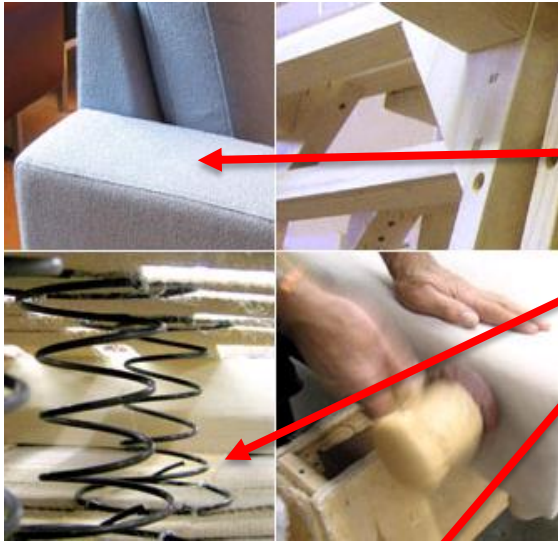
–Chemicals undergoing current screening in Arctic\*:

- Perfluorocarboxilates (PFCAs)
- Perfluoroalkylsulphonates (PFSAAs)
- Fluorotelomer alcohols (FTOHs)
- Perfluorosulphonamido-alcohols (PFTOHs)
- Siloxanes
- Organophosphate flame retardants

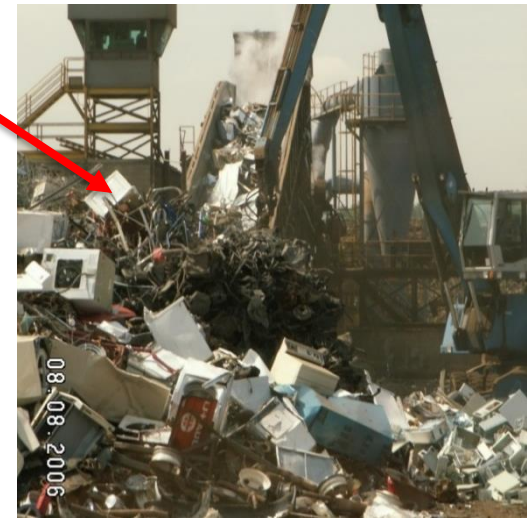
\*Northern Contaminants Program



# What if we have emerging contaminants in...



- Upholstery
- Mattresses
- Electronics
- White goods
- Vehicle parts
- Auto shredder residue



Source: Alex Stone, WA Department of Ecology

# Unique Challenges with Northern Dumpsites / Landfills

## Landfills vs. Dumpsites...really???

- Landfills: geomembrane layers, leachate and gas collection systems
- Dumpsites: unorganized areas of rubbish collection
- Location, socioeconomic conditions, disposal methods, age, sampling and sorting procedures
- Evolution and changes in leachate composition driven by pH, O<sub>2</sub> and temperature to waste layers

# Leachate Formation in a Northern Landfill/Dumpsite

- Leachate generation
- Typical contaminants
- No material completely impervious to leachate
- Improper leachate control may not be noticed until years later
- Water from precipitation, snow and waste percolates through rubbish layers
- Metals, organic compounds, (i.e. carcinogens, PFOS, EDCs, etc.), PBDEs, and dissolved organic matter

# Challenges in the Canadian North

- If contaminant is detected in the Arctic, automatically *presume* it is persistent, subject to long range transport and likely bioaccumulative
- Environmental pollution trends
  - Many declined in Arctic over past 20 years
  - Others influenced with no visible trend
- Criteria met for definition of persistent organic pollutants (POPs)
- Extent and degree of contamination poorly understood
- Release of contaminants
- Contaminants stored in: permafrost, sea ice, glaciers

## Challenges in the Canadian North

- Problems with landfills (dumpsites) across Canada
- Canadian North (i.e. north of 60° N)
- Most dumpsites unlined, some adjacent to water bodies, unauthorized dumping and/or uncontrolled access
- Northern dumpsites contain different waste streams than southern Canada
- Need to know nature and presence of emerging contaminant problem at dumpsites across the Canadian North

# Emerging Contaminants in the Canadian Arctic

## –Background on current POPs

- Legacy POPs – 23 (historical use)
- Bioaccumulate, persistent, toxic, long range transport

## –Emerging chemicals

- Many hazardous chemicals still being used commercially
- Regulated nationally/internationally: BFRs, PFOAs, current used pesticides, heavy metals (Hg, coal burning, accidents)

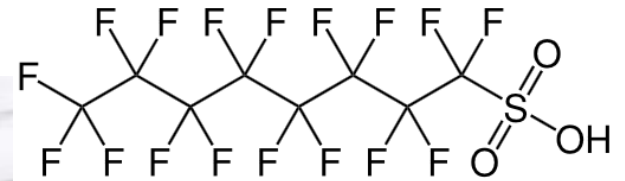
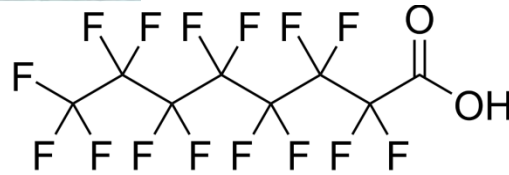


# What do these products have in common?





OR these...???

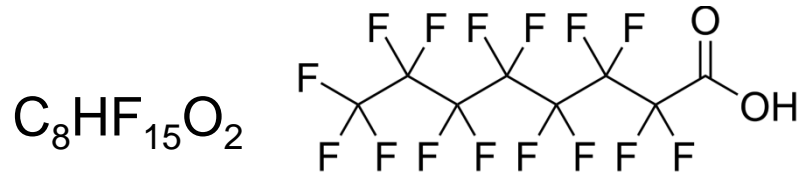


## Perfluorinated Compounds (PFCs)

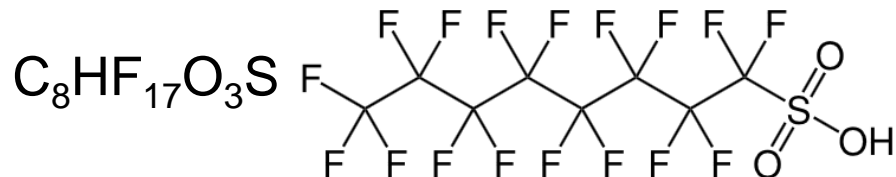
- Perfluorooctane sulphonate salts (PFOS) – used mostly in fire fighting foams
- Perfluorooctanoic Acid (PFOA) – used as emulsifier to produce fluoropolymers (PTFE or Teflon)
- Persistent, non degradable, toxic, bioaccumulate
- Uses: surface treatments, stain repellent, coating surfaces (non-stick cookware, electronic components), fire-fighting foams, waxes, food packaging

# What are PFAS Compounds?

- Synthetic class of compounds used in manufacturing fluoropolymers
  - PFOA – perfluorooctanoic acid and its principle salts, manufactured from 1947-present<sup>1</sup>, 8 manufacturers phased out production by 2015



- PFOS – perfluorooctane sulfonate, manufactured from 1949-2002<sup>2</sup>



- Used in the manufacturing of many articles of commerce



## Perfluorinated Compounds (PFCs)

- Fully fluorinated hydrophobic linear carbon chain attached to 1 or more hydrophilic heads
- WWTP: non-ionic PFCs transform into PFOS and PFOA, with minimal removal efficiency
- Sewage sludge and hence crop uptake
- No known degradation; thus it is one of the most persistent chemicals
- Long half lives - years



# Uncontrolled landfill





## Uncontrolled landfill cont'd...





# Uncontrolled landfill cont'd...





## Uncontrolled landfill cont'd...





**What Comes Next?**

## What Comes Next?

- Meta study to determine effectiveness of organics diversion ban in landfills across country
- Blanket organics ban at all landfills/dumpsites in Canada
- There may not be a facility in which to compare leachate strength with consideration to organics removal (i.e. apples to apples comparison)
  - Leachate variable over time
  - Landfill cells not ‘separate’ as waste can be filled over older cells

# Questions and Thank You!

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